



Standard Test Method for Determination of the Individual Volatile Organic Compounds (VOCs) in Air-Dry Coatings by Gas Chromatography¹

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1. Scope

1.1 This test method is for the determination of the weight percent of individual volatile organic compounds in air-dry coatings.

1.2 The method can be used to determine the weight fraction VOC content of waterborne coatings in which the material VOC content is below 5 weight percent. The method has been used successfully with higher VOC content waterborne coatings and with solvent-borne coatings (Note 1).

1.3 The method may also be used to measure the exempt volatile organic compound content (acetone, methyl acetate, *t*-butyl acetate and *p*-chlorobenzotrifluoride) of waterborne and solvent-borne coatings. The methodology is virtually identical to that used in Test Method D6133 which, as written, is specific for only exempt volatile compounds.

1.4 Volatile compounds that are present at the 0.005 weight percent level or greater can be determined.

NOTE 1—This test method may be used for the VOC analysis of coatings containing silanes, siloxanes, and silane-siloxane blends. The test method is not suitable for the analysis of coatings that cure by chemical reaction (this includes two-component coatings and coatings which cure when heated) because dilution with a solvent would impede the chemical reaction required for these types of coatings. This test method measures the VOC weight fraction of air-dry coatings directly as opposed to the methods of Practice D3960 which measure the VOC weight fraction of air-dry waterborne coatings indirectly. A direct measurement of VOC weight fraction, particularly in low VOC content waterborne coatings, generally gives better precision. California Polytechnic State University carried out an extensive study for the California Air Resources Board comparing the precision of the direct method with the indirect method (CARB Standard Agreement No. 04.329).² This study may be used to decide if the present method or the methods of Practice D3960 are preferred as an analysis method for obtaining the best possible precision for measuring the mass-based VOC content of a specific coating.

1.5 The values stated in SI units are to be regarded as standard. No other units of measurement are included in this standard.

¹ This test method is under the jurisdiction of ASTM Committee D01 on Paint and Related Coatings, Materials, and Applications and is the direct responsibility of Subcommittee D01.21 on Chemical Analysis of Paints and Paint Materials.

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² Detailed results of this study may be found at http://www.arb.ca.gov/coatings/arch/Final_Report_6_11_09.pdf.

1.6 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.*

2. Referenced Documents

2.1 ASTM Standards:³

D2369 Test Method for Volatile Content of Coatings

D3792 Test Method for Water Content of Coatings by Direct Injection Into a Gas Chromatograph

D3960 Practice for Determining Volatile Organic Compound (VOC) Content of Paints and Related Coatings

D4017 Test Method for Water in Paints and Paint Materials by Karl Fischer Method

D6133 Test Method for Acetone, *p*-Chlorobenzotrifluoride, Methyl Acetate or *t*-Butyl Acetate Content of Solventborne and Waterborne Paints, Coatings, Resins, and Raw Materials by Direct Injection Into a Gas Chromatograph

E177 Practice for Use of the Terms Precision and Bias in ASTM Test Methods

E691 Practice for Conducting an Interlaboratory Study to Determine the Precision of a Test Method

3. Terminology

3.1 Abbreviations:

3.1.1 EGDE—ethylene glycol diethyl ether

3.1.2 DB—2-(2-butoxyethoxy)ethanol; Butyl Carbitol;⁴ diethylene glycol monobutyl ether

3.1.3 EB—2-butoxyethanol; Butyl Cellosolve;⁵ ethylene glycol monobutyl ether

3.1.4 EG—ethylene glycol

3.1.5 FID—flame ionization detector

3.1.6 F-VOC—formulation data calculated volatile organic compound in g/(L-water)

3.1.7 GC—gas chromatograph

3.1.8 PG—propylene glycol

³ For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.

⁴ Butyl Carbitol is a registered trademark of The Dow Chemical Company.

⁵ Butyl Cellosolve is a registered trademark of The Dow Chemical Company.

- 3.1.9 % *RSD*—percent relative standard deviation
- 3.1.10 *SPME*—solid phase microextraction
- 3.1.11 *SPDE*—solid phase dynamic extraction
- 3.1.12 *Std Dev*—standard deviation
- 3.1.13 *TX*—2,2,4-trimethylpentane-1,3-diol, monoisobutyrate
- 3.1.14 *VOC*—volatile organic compound
- 3.1.15 *X-VOC*—experimental volatile organic compound in g/(L-water)

4. Summary of Test Method

4.1 A known weight of coating is dispersed in methanol or tetrahydrofuran (THF) internally standardized, and analyzed by capillary gas chromatography to give a speciated composition of the volatile organic compounds and exempt organic compounds, if any, present in the coating. Summation of the individual volatile organic compound weight fractions gives the total VOC content of the coating measured in weight percent (Note 2).

NOTE 2—Methanol should be used as a solvent for all waterborne coatings. THF should be used for solventborne coatings. Acetone may also be used for solventborne coatings but should not be used for waterborne coatings because it may react with ammonia and amines which are frequently found in waterborne coatings. Using the provisions of Practice D3960, the VOC content of coatings measured in g/L minus water, or other units, may be determined. Since the determination of weight percent VOC in the present method is by direct measurement, either the water fraction (Test Method D3792 or Test Method D4017) or the nonvolatile fraction (Test Method D2369) may be determined indirectly in the application of Practice D3960. Since precision is better for the determination of the nonvolatile content, this is the preferred method for the indirect calculation of water content in this test method. The equations for calculating coating VOC content when no exempt volatile compounds are present are:

$$VOC = \frac{f_{VOC}(D_P)}{1 - [(1 - f_{NV} - f_{VOC})(D_P/D_W)]} \quad (1)$$

or

$$VOC = \frac{f_{VOC}(D_P)}{1 - [f_W(D_P/D_W)]} \quad (2)$$

where:

D_P , f_{NV} , f_{VOC} , and f_W = coating density, nonvolatile fraction, VOC fraction, and water fraction, respectively.

4.2 Direct GC/FID, GC/MS and solid phase microextraction / gas chromatography (SPME/GC) of the coating may be used to facilitate identification of the volatile compounds present in a coating. Table X1.1 lists the GC retention times for the volatile compounds which may be found in low VOC content air-dry coatings and several possible internal standards, ordinarily not present in coatings, which may be used (Note 3).

NOTE 3—The analyst should consult MSDS and product data sheets for possible information regarding solvents which may be present in a particular coating. Retention times given in X1.1 must be verified for each individual instrument.

5. Significance and Use

5.1 In using Practice D3960 to measure the VOC content of waterborne coatings, precision tends to be poor for low VOC

content air-dry coatings when the VOC weight fraction is determined indirectly. The present method first identifies and then quantifies the weight fraction of individual VOCs directly in air-dry coatings. The total VOC weight fraction can be obtained by adding the individual weight fraction values (Note 4).

NOTE 4—The present method may be used to speciate solvent-borne air-dry coatings. However, since these normally contain high, and often complex, quantities of solvent, precision tends to be better using the methods contained in Practice D3960, where the VOC fraction is determined by a direct weight loss determination.

5.2 SPME/GC makes it possible to identify very low levels of volatile compounds in a coating and could serve to make it possible to identify the presence of hazardous air pollutants (HAPs).

6. Apparatus

6.1 *Gas Chromatograph, FID Detection or Mass Spectrometry Detection with Electronic Data Acquisition System*—Any capillary gas chromatograph equipped with a flame ionization detector or mass spectrometer and temperature programming capability may be used. Electronic flow control, which gives a constant carrier gas flow, is highly recommended.

6.2 *Standard GC/FID and GC/MS Instrument Conditions:*

6.2.1 See Table 1.

7. Reagents and Materials

7.1 *Purity of Reagents*—Reagent grade chemicals shall be used in all tests. Unless otherwise indicated, all reagents shall conform to the available specifications of the Committee on Analytical Reagents of the American Chemical Society. Other grades may be used, provided it is first ascertained that the reagent is of sufficiently high purity to permit its use without lessening the accuracy of the determination.

7.2 *Carrier Gas*, helium of 99.995 % or higher purity.

7.3 *Tetrahydrofuran (THF)*, HPLC grade.

7.4 *Methanol*, HPLC grade.

7.5 Possible internal standards: *1-Propanol*, *p*-fluorotoluene, *cyclohexanol*, *p*-chlorotoluene, *ethylene glycol diethylether (EGDE)*.

7.6 *Fluorocarbon-faced Septum Vials*, 20 mL and 40 mL capacity.

7.7 *Ceramic Beads*, 0.5–1.0 mm diameter.

8. Column Conditioning

8.1 The capillary columns should be conditioned according to the manufacturer's recommendation. The columns may then be used indefinitely without further conditioning.

9. Preparation of Standards

9.1 Prepare a stock mixture of ethylene glycol (EG), propylene glycol (PG), ethylene glycol monobutyl ether (EB), ethylene glycol diethyl ether (EGDE) [or other suitable internal standard], diethylene glycol monobutyl ether (DB), and 2,2,4-trimethylpentane-1,3-diol monoisobutyrate (TX) by weighing one or two grams of each into an appropriate vial. The weight of each component should be approximately the same and determined to 0.1 mg. Mix the contents.

TABLE 1 Standard FID Instrument Conditions

Instrument Conditions	FID	GC/MS
Detector	Flame ionization	70 eV electron impact mass spectrometer
Columns	Primary column: 30 by 0.25 mm 5 % phenyl/95 % methyl siloxane (PMPS), ^A 1.0 µm film thickness Confirmatory Columns: 30 by 0.25 mm polydimethylsiloxane (PDMS), 0.25 µm film thickness; 30 by 0.25 mm Carbowax (CW), 0.25 µm film thickness.	Primary column: 30 by 0.25 mm % phenyl/95 % methyl siloxane (PMPS), ^A 1.0 µm film thickness Confirmatory Columns: 30 by 0.25 mm polydimethylsiloxane (PDMS), 0.25 µm film thickness; 30 by 0.25 mm Carbowax (CW), 0.25 µm film thickness.
Carrier Gas	Helium	Helium
Flow Rate	1.0 mL per min, constant flow (24.9 cm/s at 40°)	1.0 mL per min, constant flow (24.9 cm/s at 40°)
Split Ratio	50 to 1	50 to 1
Temperatures, °C, Primary Column		
Inlet	260°	260°
Detector	270°	270°
Initial	50° for 4 min	50° for 4 min
Rate	20° per min to 250°, hold 6 min (total run time = 20 min)	20° per min to 250°, hold 6 min (total run time = 20 min)
Temperatures, °C, Confirmatory Columns		
Inlet	260°	260°
Detector	270°	270°
Initial	40° for 4 min	40° for 4 min
Rate	10° per min to 250°, hold 25min	10° per min to 250°, hold 25min
Source		230°
Quadrupole		230°
Transfer Line		260°
Scanning Parameters		Mass 29–400 amu

^A Some coatings may contain high-boiling components which elute from the GC capillary column after the specified run time of 20 minutes. It is advisable, therefore, to bake out the column in these cases. The column designated as PMPS is commercially available from several vendors by the following designations: DB-5, SPB-5, HP-5, AT-5, CP Sil 8CB, RTX-5, BP-5. The column designated as PDMS is available by the designations DB-1, SPB-1, HP-1, AT-1, CP Sil 5 CB, RTX-1. The column designated as Carbowax is available by the designations Supelcowax 10, DB-Wax, HP-Wax, AT-Wax, CP-Wax 52 CB, RTX-Wax, BP-20.

9.2 Transfer approximately 100 µL of the stock mixture to a septum-capped vial containing 10 mL of THF or methanol and mix the contents (**Note 5**). This solution will contain each of the known analytes at a concentration of approximately 2 mg/mL.

NOTE 5—The solvents EG, PG, EB, DB TX are widely used in the manufacture of waterborne air-dry coatings and may be expected as highly probable components of these coatings.

9.3 Chromatograph the solution in 9.2 by injecting 1 µL into the PMPS column using the chromatographic conditions given in 6.2. Calculate the relative response factors for each of the analytes relative to the EGDE or other suitable internal standard using the relationship:

$$RF = \frac{AA * MI}{AI * MA} \quad (3)$$

where:

RF = relative response factor,

AA = area of analyte,

MI = weight of internal standard (from 9.1),

AI = area of internal standard, and

MA = weight of analyte (from 9.1).

10. Paint Analysis

10.1 *Analysis of Air-dry Solvent-borne and Waterborne Coatings by GC/FID or GC/MS:*

10.1.1 Prepare duplicate samples by pipetting 10 mL of methanol (waterborne coatings) or THF (solventborne coatings) into a vial containing 3 to 5 g of ceramic beads and close with a fluorocarbon-faced septum cap. Using a dedicated glass syringe (25 or 50 microliter capacity), add 10 µL of EGDE or other internal standard and weigh to at least 0.1 mg. This

solution must be analyzed by GC to determine if there are peaks that result from it rather than from the paint sample that is prepared in 10.1.2.

10.1.2 Pipette 10 mL of methanol or THF into a 20 or 40 mL vial containing 3 to 5 g ceramic beads and close with a fluorocarbon-faced septum cap. Using a disposable 1 mL syringe, add approximately 0.6 to 0.8 g of the well-mixed paint through the septum cap and weigh to 0.1 mg (Note 6). Using the dedicated syringe, add 10 microliters of pure EGDE (or other internal standard) through the septum and weigh the amount added to at least 0.1 mg. Mix the contents vigorously by shaking for 1 min. Let the vial stand to permit pigments, if any, to settle.

NOTE 6—The paint should be drawn into the syringe without an attached syringe needle. Excess paint is wiped from the syringe and the needle is then attached for paint transfer. The mass of the paint may be determined by either the difference in the weight of the filled and empty syringe or by the difference in the weight of the vial before and after adding paint.

10.1.3 Chromatograph the solution in 10.1.2 by injecting 1 µL into the PMPS capillary column using the standard conditions described in 6.2. Adjust the split ratio to give well-defined chromatographic peaks. Identify the volatile compounds which elute over a 20 minute run time. An internal marker, methyl palmitate (BP = 338°C) should elute at approximately 18.4 minutes. Calculate the weight fraction of each peak using the relationship:

$$\%X = \frac{(AA)(MI)(100)}{(AI)(RF)(MC)} \quad (4)$$

where:

X = one of several possible volatile compounds in the coating,

RF = relative response factor of compound X ,

AA = peak area of compound X ,

MI = weight of internal standard,

AI = peak area of internal standard, and

MC = weight of coating.

NOTE 7—If volatile compounds other than those in the standard (9.1) are present in the coating, the identity should be confirmed by FID retention time comparison with authentic material or by GC/MS and the relative response factor should be determined as outlined in 9.1-9.3. Commercial 2,2,4-trimethylpentane-1,3-diol monoisobutyrate (TX) may contain small amounts of 2,2,4-trimethylpentane-1,3-diol which elutes approximately 0.5 minutes before butyl carbitol and 2,2,4-trimethylpentane-1,3-diol diisobutyrate (TXIB) which elutes approximately 1.5 minutes after 2,2,4-trimethylpentane-1,3-diol monoisobutyrate (TX). Acetone and isopropyl alcohol have nearly the same retention time on a PMPS column and if either is found, their identities should be confirmed and quantitated on a Carbowax⁶ column or by using GC/MS. Isobutyl alcohol coelutes with the solvent (THF) and must be determined on a different column (Carbowax) or using a different solvent (methanol). SPME, SPDE and static headspace analysis are especially useful techniques for confirming that decomposition products are not being observed. Small quantities (up to 0.5 %) of acetic acid are sometimes found in coatings containing vinyl acetate resins. The acetic acid is formed as a decomposition product in the GC inlet and should not be counted as a VOC. Some coatings contain additives (for example, carbamate ester

biocides) that may give decomposition products in the hot inlet of the gas chromatograph. If decomposition products are suspected, a convenient procedure for determining this is to analyze the coating by static headspace gas chromatography. In using static headspace, a large sample of the coating (15 to 20 g) is internally standardized with 10 mg/g of EGDE, approximately 5 mL of ceramic beads are added, and manually mixed by shaking until the paint/internal standard mixture is homogeneous. The static headspace procedure is carried out on 40 to 60 mg of the internally standardized coating using a 20 mL crimp-cap headspace vial. Static headspace conditions are: Oven, 20 to 30 minutes at 130°C; Loop, 150°C; Transfer Line, 150°C. Chromatographic Conditions: as described in 6.2. Alternatively, if static headspace, SPME, or SPDE capability are not available, the analysis can be done using a lower inlet temperature as long as the selected temperature is high enough to fully volatilize the suspect compounds. Cool on-column injection can also be used to determine if a compound is being generated via thermal decomposition in the hot GC inlet.

10.2 Overlapping Chromatographic Peaks:

10.2.1 A number of VOCs in solvent-borne paints containing commercial xylene tend to overlap. These include propylene glycol monomethyl ether acetate/ethylbenzene and butoxyethanol/o-xylene. Resolution can generally be obtained by simply changing the chromatographic heating rate.

10.3 Coatings Containing Silanes, Siloxanes and Silane-Siloxane Blends:

10.3.1 If the coating contains silanes, siloxanes, and silane-siloxane blends, approximately 50 mg of solid p-toluenesulfonic acid should be added to the solution in 10.1.2 thirty minutes prior to gas chromatography. The p-toluenesulfonic acid catalyzes the hydrolysis of alkoxy silanes to free alcohol (usually ethanol).

11. Reporting Results

11.1 Prepare a table (as indicated below) which contains information on each of the VOC species found. Do not include any exempt volatile organic compounds in this table.

VOC Found	GC Retention Time	Weight % Found

Total weight percent of all speciated VOCs =

11.2 List VOCs that are not identified as unknown (UK) and use the relative response factor of 2,2,4-trimethylpentane-1,3-diol monoisobutyrate to calculate the weight % for unknown VOCs. List any exempt compounds that may be present separately from those listed in the VOC table.

12. Alternate Identification Methods

12.1 The use of GC/MS for volatile compound identification is highly desirable even when quantitation is carried out by GC/FID. A convenient procedure is to sample the headspace of the coating using an SPME or SPDE followed by thermal desorption onto any standard capillary column and subsequent mass spectral identification. This technique is especially valuable for identifying oxygenates, aromatics and other volatile organic compounds.

13. Precision and Bias

13.1 An interlaboratory study of total weight percent VOC was conducted in accordance with Practice E691 in seven

⁶ Carbowax is a registered trademark of The Dow Chemical Company.

TABLE 2 Precision Statistics for Low VOC Content Waterborne Air-Dry Coatings Analyzed by Gas Chromatography

Material	Total VOC in Weight Percent, Average	Repeatability Standard Deviation	Reproducibility Standard Deviation	Repeatability Limit	Reproducibility Limit
A	0.25	0.009	0.020	0.03	0.06
B	2.85	0.058	0.125	0.16	0.35
C	4.04	0.087	0.187	0.24	0.52
D	4.49	0.145	0.314	0.41	0.88
E	4.50	0.100	0.215	0.28	0.60

laboratories with five materials, with each laboratory obtaining three test results for each material. Five commercial waterborne coatings, a primer, a flat, a satin, a semi-gloss and a gloss, ranging in weight percent VOC content from 0.25 to 4.50, were analyzed. Each of the laboratories analyzed the coatings three times according to protocols specified in Practice E691 (see Note 8).

13.2 Precision statistics were calculated for the total weight percent VOC found in each of the five coatings and are presented in Table 2. The terms repeatability limit and reproducibility limit are used as specified in Practice E177.

13.3 95 % *Repeatability Limit (within laboratory)*—The within-laboratory coefficient of variation is 2.7 % relative. The 95 % confidence limit for the difference between two such averages is 7.5 % of the test result.

13.4 95 % *Reproducibility Limit (between laboratories)*—The between-laboratory coefficient of variation is 5.8 % relative. The 95 % confidence limit for the difference between two such averages is 16.2 % of the test result.

13.5 *Bias*—Bias has not been determined.

NOTE 8—The precision statistics determined for this test method are applicable only to gas chromatography using flame ionization detection. Statistics using mass spectral detection have not been determined.

14. Keywords

14.1 exempt organic compounds; gas chromatography; speciation; VOC (volatile organic compound); waterborne coatings

APPENDIX

(Nonmandatory Information)

X1. RELATIVE RESPONSE FACTORS AND RETENTION TIMES – PRECISION STATISTICS

X1.1 Precision statistics were calculated for each of the individual VOCs found in the five coatings analyzed in the interlaboratory study and are presented in Table X1.2. The

experimental VOC content (X-VOC) is compared with the theoretical VOC (F-VOC) obtained from formulation data.

TABLE X1.1 Retention Times (in Minutes) and FID Relative Response Factors (RRF) of Possible Volatile Organic Compounds and Exempt Compounds in Waterborne Air-Dry Coatings Using Chromatographic FID Conditions Described in 6.2

Compound	RRF	PMPS	Carbowax	PDMS
methanol	0.66	2.35	5.16	1.97
ethanol	1.00	2.75	5.85	2.08
acetone	1.02	3.12	3.93	2.21
2-propanol	1.02	3.13	5.71	2.19
t-butyl alcohol		3.42		
methyl acetate	0.60	3.59	4.1	2.31
1-propanol		3.82	7.91	2.44
vinyl acetate		4.40		
methyl ethyl ketone	1.25	4.74	5.24	2.7
2-butanol	1.18	4.78	7.63	2.72
ethyl acetate	0.83	5.05		
tetrahydrofuran (THF)		5.31	4.65	3.05
isobutyl alcohol	1.62	5.31	8.94	2.97
1-butanol	1.59	5.97	9.92	3.39
propylene glycol monomethyl ether	0.79	6.18	9.66	3.62
ethylene glycol	0.55	6.42	16.94	3.91
ethyl acrylate		6.48		
t-butyl acetate	1.22	6.65	5.31	4.18
methyl methacrylate		6.74		
2-ethoxyethanol	0.70	6.83		
methyl pivalate		7.00		
propylene glycol	0.73	7.18	16.48	4.8
AMP-95	0.84	7.46		
toluene	2.07	7.82	8.02	5.44
p-fluorotoluene		8.03		
ethylene glycol diethyl ether (EGDE)	1.00	8.12		
2-propoxyethanol (EP)	1.00	8.27	12.58	6.41
butyl acetate	1.16	8.33		
propylene glycol monopropyl ether (PnP)	1.01	8.72	11.65	7.32
diacetone alcohol		8.73		
furfuryl alcohol	0.93	8.88		
ethylbenzene	2.04	8.92	10.87	7.75
parachlorobenzotrifluoride	1.02	8.93		
propylene glycol, mono t-butyl ether	1.16	8.96	12.58	7.67
propylene glycol, methyl ether acetate	0.86	8.99		
p-xylene	2.17	9.19	9.72	7.6
m-xylene	2.17	9.19	9.87	7.81
butyl ether		9.20	9.99	7.81
2-heptanone	1.46	9.32		
cyclohexanol		9.32		
butyl acrylate		9.32		
butyl propionate		9.45		
o-xylene	2.17	9.47		
2-butoxyethanol (EB)	1.11	9.51	10.78	8.28
hexylene glycol		9.62	14.05	8.6
diethylene glycol monomethyl ether	0.70	9.81	16.89	
propylene glycol monobutyl ether (PnB)	1.09	9.87	16.52	
diethylene glycol diethyl ether	0.68	9.88		
PnB (minor component)		10.05	13.15	9.41
glycerin	0.65	10.12		
diethylene glycol	0.57	10.14		
p-chlorotoluene		10.17	13.97	9.66
butyl methacrylate		10.20		
dipropylene glycol monomethyl ether (DPM)	0.80	10.46, 10.51, 10.71		
diethylene glycol monoethyl ether	0.69	10.49		
p-cymene	2.13	10.84	15.62	10.67
benzyl alcohol	1.65	10.89		
N-methylpyrrolidinone	0.72	10.96	12.18	10.94
ethylene glycol butyl ether acetate		11.25	17.73	
diethylene glycol monopropyl ether (DP)		11.34	14.8	
diethylene glycol monopropyl ether (DPnP)	0.76	11.79, 11.82		11.85

TABLE X1.1 *Continued*

Compound	RRF	PMPS	Carbowax	PDMS
2,2,4-trimethylpentane-1,3-diol				
diethylene glycol monobutyl ether (DB)	0.93	12.12		
naphthalene	2.03	12.43		
2-(2-ethylhexyl)ethanol		12.51		
triethylene glycol	0.58	12.53		
dipropylene glycol monobutyl ether (DPnB)	1.04	12.53, 12.58	18.86	13.41
propylene glycol monophenyl ether	1.14	12.67		
2,2,4-trimethylpentane-1,3-diol monoisobutyrate (TX)	1.27	13.62, 1374		
tetraethylent glycol	0.59	14.54		
2,2,4-trimethylpentane-1,3-diol, diisobutyrate (TXIB)	1.17	15.15		
2-ethylhexyl benzoate	1.39	16.05		
methyl palmitate		18.40		
dibutyl phthalate	1.16	18.50		
Mineral Spirits	2.23	9 to 11		
Aromatic100	2.10	9.2 to 10.6		

TABLE X1.2 Precision Statistics for Individual VOCs Found in the Round Robin of Coatings A-E

Paint		EG %	PG %	EB %	DB %	TX %	Total VOC, %	X-VOC ^A	F-VOC ^A
A	Average					0.25	0.25	10	11
	Std Dev					0.03	0.03	1.2	
	% RSD					11.93	11.93	11.8	
B	Average		1.53		0.14	1.19	2.85	92	92
	Std Dev		0.14		0.01	0.10	0.11	3.1	
	% RSD		9.18		9.13	8.13	3.76	3.4	
C	Average		2.49			1.55	4.04	145	147
	Std Dev		0.11			0.14	0.16	5.0	
	% RSD		4.61			8.76	4.00	3.4	
D	Average	2.31		1.04	0.09	1.05	4.49	113	121
	Std Dev	0.32		0.07	0.01	0.11	0.23	5.2	
	% RSD	13.65		6.73	8.51	10.17	5.09	4.6	
E	Average		2.40		0.64	1.45	4.50	142	138
	Std Dev		0.11		0.05	0.15	0.15	4.0	
	% RSD		4.66		8.23	9.99	3.29	2.8	

^A Measured in grams per litre minus water.

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